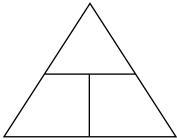
Date: <sub>-</sub>		Period:	Earth Science
		Packet: Min	erals
CLASS	S NOTE	S	
•	Rock -	any naturally formed solid that is part of Earth	or any other celestial object
	•	Minerals are the	_ needed to form the different types of rocks
•	<u>Minera</u>	al - most geologist define a "mineral" as:	
	1.		
	2.		
	3.		
	4.		
	5.		
•	Minera	als form in the following manners:	
	1.	Precipitation from evaporating seawater	
	2.	Crystallization around cooling magma	
	3.	Under extreme heat and pressure [recrystalliz	ration]
	4.	From hydrothermal solutions flowing through	ground cracks
•	Most r	ock forming minerals are silicates that result in	a tetrahedron shape
	•	Four-sided units of 4	and 1

•	Physical and chemical properties of a minerals are determined by the:			
•	Each r	mineral has a set of physical and chemical properties that can be used to identify the sample		
•	The fol	llowing methods are used to classify minerals:		
	1.	Color - a visual attribute of an object based on perception		
		<ul> <li>One of the most obvious, but not the most reliable</li> </ul>		
		Many of the known minerals share similar colors		
	2.	Streak -		
		Weathering changes the outside color, but streak gives the true color		
	3.	<u>Luster</u> -		
		Two types of luster:		
		Metallic Luster - shines like stainless steel		
		Nonmetallic Luster - earthy or dull shine		
	4.	Density -		
		-		



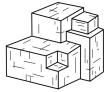
Methods used to classify minerals [continued]:

5.	Hardness -	

• Mohs Hardness Scale is used to classify hardness

Hardness	Mineral	Test	
1	Talc	Finger nail scratches easily	
2	Gypsum	Finger nail scratches	
3	Calcite	Copper penny scratches	
4	Fluorite	Steel knife scratches easily	
5	Apatite	Steel knife scratches	
6	Feldspar	Steel knife will not scratches	
7	Quartz	Will scratch glass and steel	
8	Topaz	Harder then any common mineral	
9	Corundum	Scratches topaz	
10	Diamond	Hardest mineral	

- 6. Cleavage \_\_\_\_
  - Example: \_\_\_\_\_









•	Metho	ds used to classify minerals [continued]:
	7.	Fracture -
		• Example:
		Fracture Examples
	8.	Acid Test - when hydrochloric [HCI] acid is placed on a mineral it can effervesce [bubble]
		• Example:
	9.	Magnetism - when a mineral is attracted to a magnet
		• Example:
	10	Example:  Example:
	11	. <u>Taste</u> - when a mineral tastes salty
		• Example:
	12	Smell - when a mineral exhibits a distinctive smell
		• Example:
	13	<ul> <li><u>Radioactivity</u> - some minerals gives off radiation that can be measured with Geiger counter</li> <li>Example:</li> </ul>

#### PART I QUESTIONS: MULTIPLE CHOICE

- 1. Which common nonmetallic mineral has a white-yellow streak?
  - a. graphite
  - b. calcite
  - c. sulfur
  - d. dolomite
- 2. The mineral mica breaks evenly along flat sheets mainly because of its
  - a. density
  - b. chemical composition
  - c. atomic arrangement
  - d. hardness
- 3. Which mineral has nonmetallic luster, exhibits cleavage, and feels greasy?
  - a. halite
  - b. gypsum
  - c. talc
  - d. olivine
- 4. Which property is most useful in mineral identification?
  - a. size
  - b. color
  - c. texture
  - d. hardness
- 5. Certain minerals usually break along flat surfaces, while other minerals break unevenly. This characteristic is due to the
  - a. luster of the mineral
  - b. age of the mineral
  - c. force with which the mineral is broken
  - d. internal arrangement of the mineral's atoms
- 6. The physical properties of a mineral are largely due to its
  - a. internal arrangement of atoms
  - b. volume
  - c. organic composition
  - d. melting point
- 7. Which element combines with silicon to form the tetrahedral structure of the silicate minerals?
  - a. nitrogen
  - b. potassium
  - c. hydrogen
  - d. oxygen

<ul> <li>8. Quartz mineral samples are best identified by their</li> <li>a. hardness</li> <li>b. color</li> <li>c. size</li> <li>d. mass</li> </ul>
<ul> <li>9. Which property is least useful in mineral identification?</li> <li>a. streak</li> <li>b. color</li> <li>c. luster</li> <li>d. hardness</li> </ul>
<ul> <li>10. Which property of a mineral most directly results from the internal arrangement of its atoms?</li> <li>a. volume</li> <li>b. color</li> <li>c. crystal shape</li> <li>d. streak</li> </ul>
<ul> <li>11. Which common minerals fizzes when dilute hydrochloric acid [HCI] is placed on it?</li> <li>a. calcite and feldspar</li> <li>b. feldspar and quartz</li> <li>c. quartz and dolomite</li> <li>d. calcite and dolomite</li> </ul>
<ul> <li>12. Which of the following minerals has metallic luster, silver color, black streak, and contains iron?</li> <li>a. graphite</li> <li>b. galena</li> <li>c. magnetite</li> <li>d. pyrite</li> </ul>
<ul> <li>13. The minerals talc, muscovite mica, quartz, and olivine are similar because they</li> <li>a. have the same hardness</li> <li>b. are the same color</li> <li>c. contain silicon and oxygen</li> <li>d. break along cleavage planes</li> </ul>
<ul> <li>14. Which mineral is commonly mined as a source of the element lead [Pb]?</li> <li>a. galena</li> <li>b. magnetite</li> <li>c. quartz</li> <li>d. gypsum</li> </ul>
15. Which mineral will scratch fluorite, galena, and pyroxene?  a. graphite b. calcite

c. olivined. dolomite

For questions 16 through 19, use the table below that shows data for a student's collection of mineral samples A through I. For each mineral sample, the student recorded mass, volume and density. The density for rock D and the volume for rock B have been left blank.

Mineral	Mass [grams]	Volume [cm³]	Density [g/cm³]
А	82.9	34.4	2.41
В	114.2		2.68
С	144.7	63.2	2.29
D	159.4	59.7	
Е	87.7	33.1	2.65
F	59.6	21.0	2.84
G	201.1	68.4	2.94
Н	85.1	11.2	7.60
I	110.2	47.3	2.33

- 16. The approximate density of rock sample D is
  - a. 2.75 g/cm<sup>3</sup>
  - b. 3.75 g/cm<sup>3</sup>
  - c. 3.32 g/cm<sup>3</sup>
  - d. 2.67g/cm<sup>3</sup>
- 17. The approximate volume of rock sample B is
  - a. 12.6 grams
  - b. 22.0 grams
  - c. 32.5 grams
  - d. 42.6 grams
- 18. Based in the density data, what is the name of Mineral H?
  - a. Graphite
  - b. Sulfur
  - c. Galena
  - d. Quartz
- 19. The student broke rock G into two pieces. Compared to the density of the original rock, the density of one piece would most likely be?
  - a. the same
  - b. greater
  - c. less